

Spotlights: Research Expedited by HPC

High-Throughput Computational Materials Design for Catalysis and Renewable Energies

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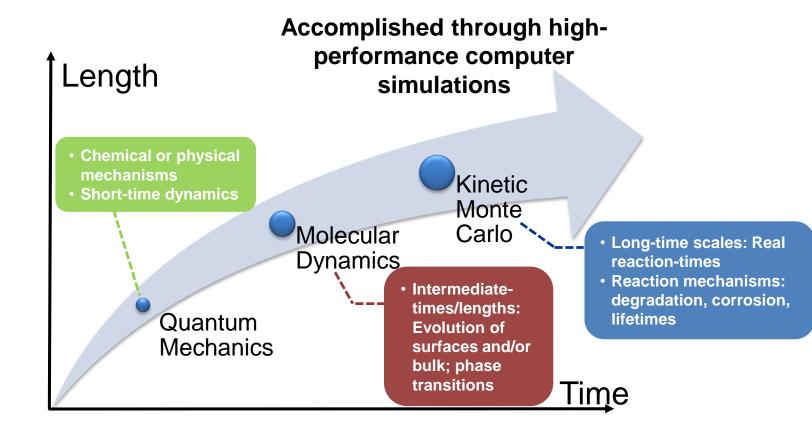
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High-Throughput Computational Materials Design

Goals: Understand/predict reaction & degradation mechanisms and dynamic evolution of materials under reaction and/or harsh environmental conditions



Ada Cluster



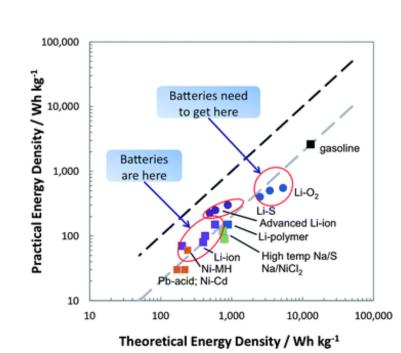


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Renewable energies (solar, wind) are intermittent



materials research crucial to develop electric vehicle applications



Batteries store chemical energy





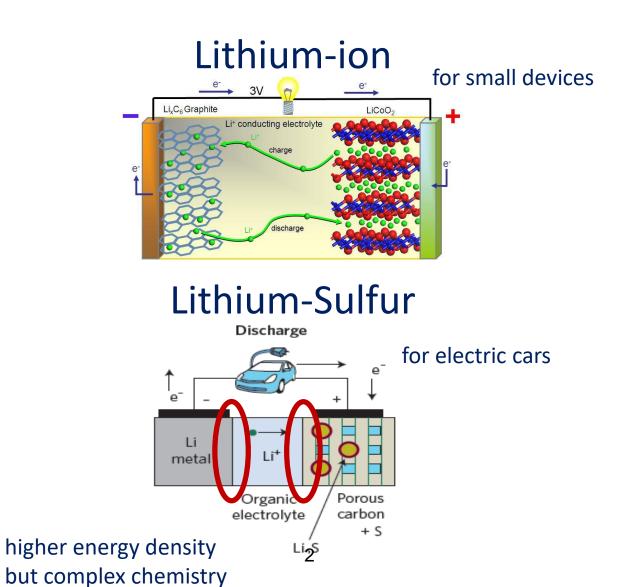
molecular level understanding is vital

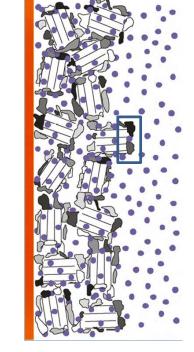
Thackeray et al., Energy Environ. Sci., 5, 7854, (2012)



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interfacial reactions generate a multicomponent solidelectrolyte interphase (SEI) film. Its properties are key for the battery lifetime

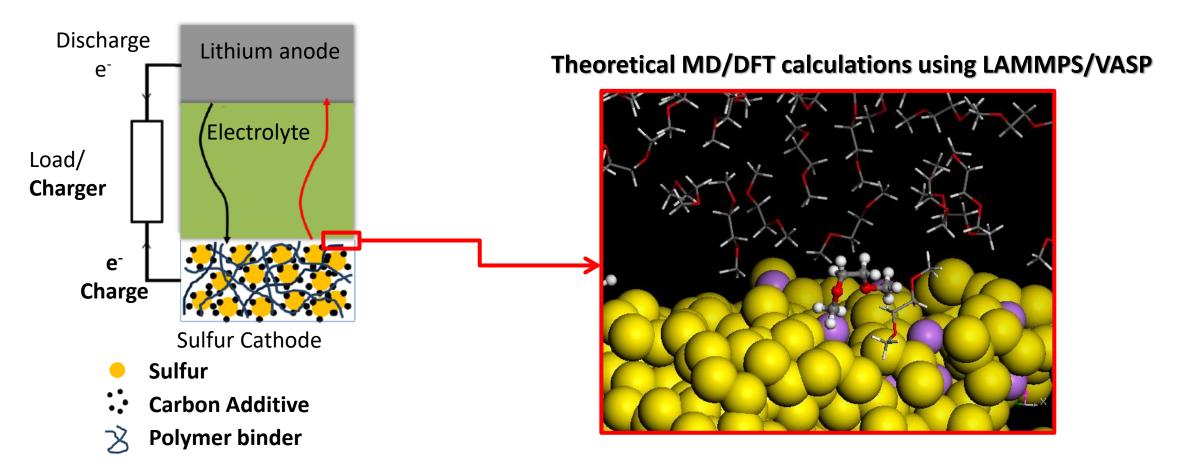
New insights obtained through DFT and AIMD simulations in high performance computers!!!

software: Gaussian 09, VASP; hardware: ADA



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Li-S batteries: Elucidation of the Cathode-Electrolyte Interfacial Chemistry

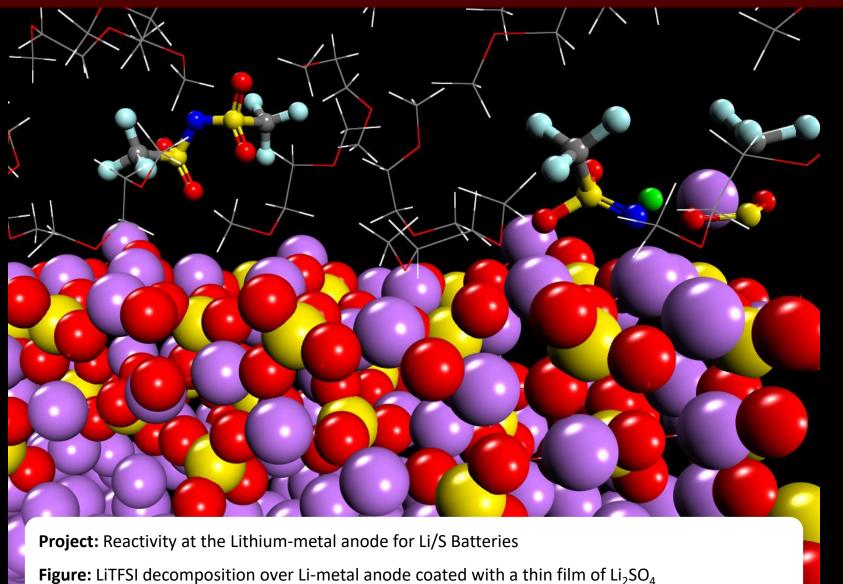


HPC Ada cluster at Texas A&M (IBM NeXtScale Cluster)



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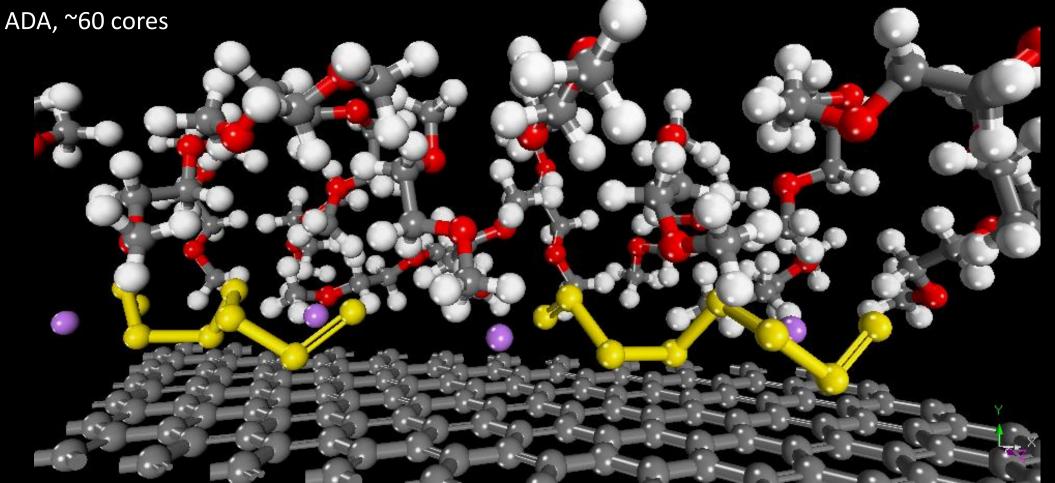




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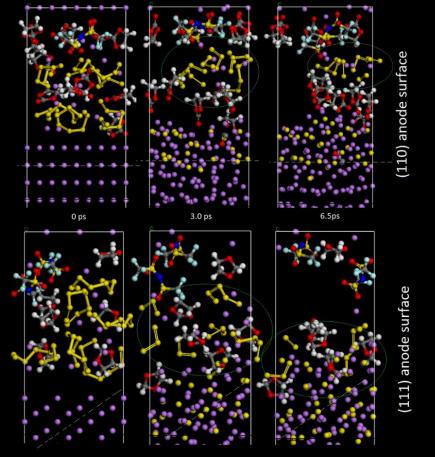
Project: Lithium Sulfur Battery: cathode and electrolyte stability and behavior **Figure:** Mixture of DME and polysulfide molecules on top of graphene layer



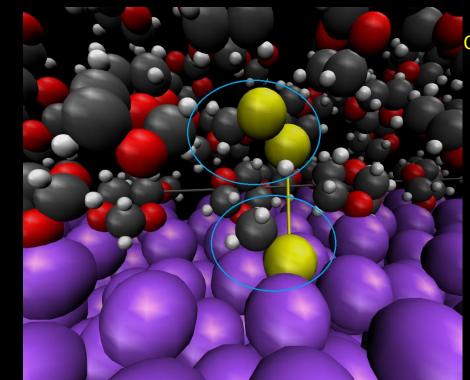


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Reactivity of PS and OS on Lithium Anode Surface of Li-S batteries



Reactivity of polysulfide (PS) decomposition on Li metal



Organosulfide (OS) radicals formed at the Li metal surface

DFT-based Studies of the impact of "Shuttle-Effect" on SEI formation





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Project: Solid-electrolyte interphase (SEI) formation and growth

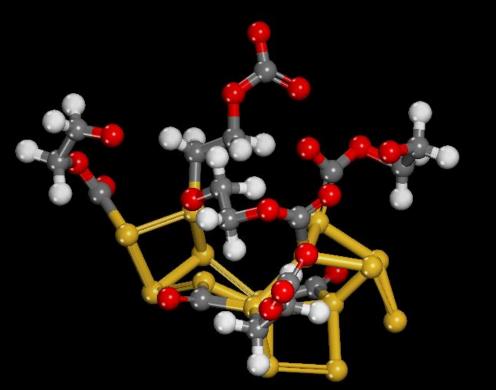
Figure: Solvent decomposition over Si anode



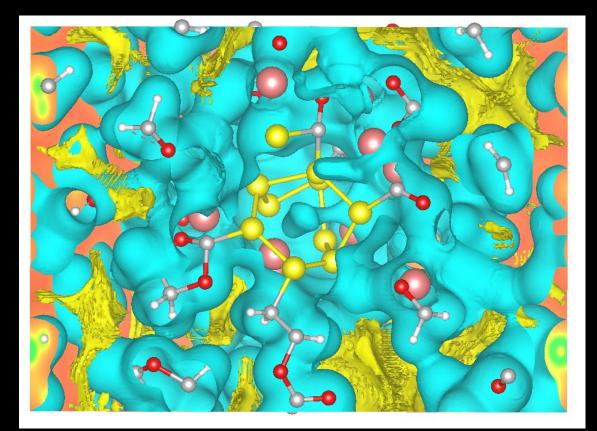
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SEI Formation on Si Nanocluster

software: VASP, hardware: ADA, ~60 cores



Initial ethylene carbonate adsorption and decomposition on Si cluster



Electrostatic potential map for Si cluster and solvent decomposition products

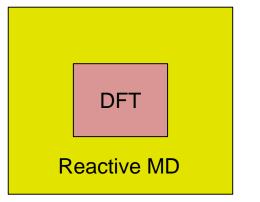




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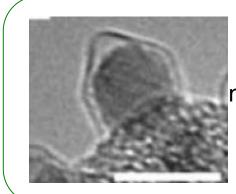
Catalyzed Synthesis of Single-Walled Carbon Nanotubes





DFT calculations Small size models: detailed investigation of interatomic interactions, most stable configurations, nature of interactions, minimum energy reaction paths

software: Gaussian09, VASP; our own MD code ; hardware: Ada, Curie, Brazos



Hypothesis:

In this catalytic process, the nanoparticle structure can act as a template to guide nanotube growth toward desired chiralities.

high resolution tunneling electron microscopy image

Reactive MD simulations

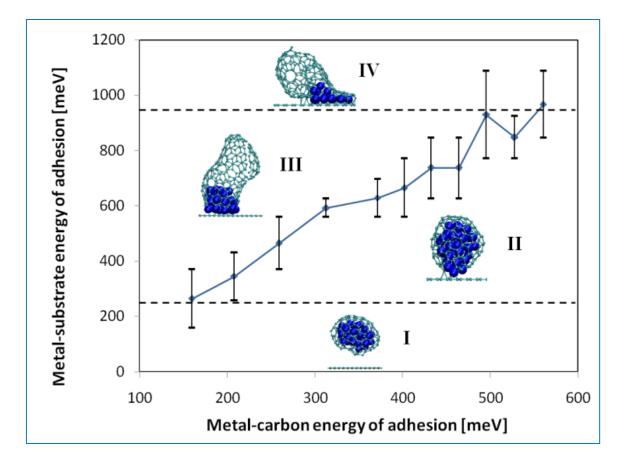
Larger size models: growth mechanism using DFTderived parameters. Temperature effects, dynamicdependent properties, cost-efficient exploration of parameter space

State and evolution of the system dictated by thermodynamic and kinetic factors



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Carbo Nanotube Growth Phase Diagrams



Phase diagrams illustrate regions where high quality (III) or defective tubes may grow (IV) or where the catalyst may become deactivated (I and II)

Accomplished through simulations in high performance computers!!!

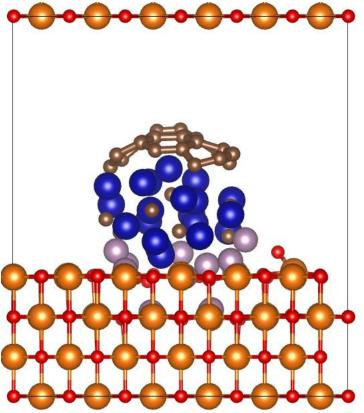
software: Reactive MD program developed in Balbuena's group hardware: EOS, Lonestar, Stampede

Burgos, Jones, Balbuena, JPCC, 118, 4808-4817, 2014

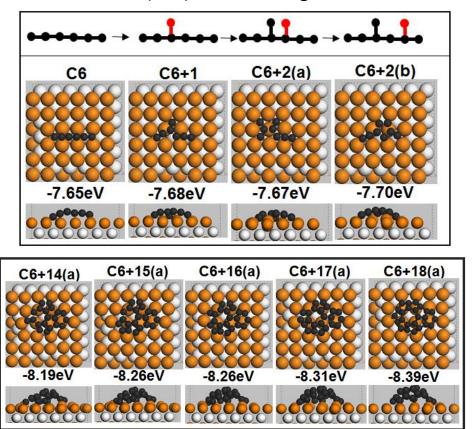


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The Study of Graphene/CNT Growth on MgO-Supported Carburized Cobalt Nanoparticles



Thorough study of the CoMoCAT Process used for growing CNT We explore pathways of growth of graphene on the Cu(100) surface using DFT

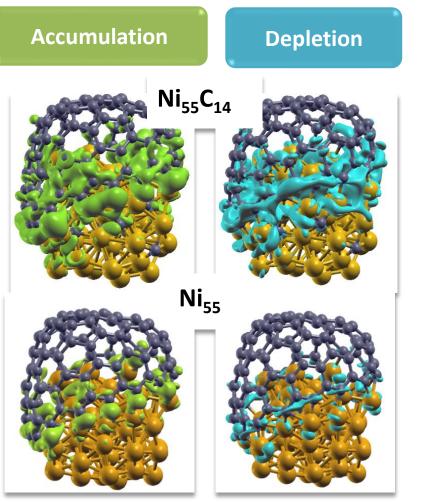


software: VASP hardware: Ada, Curie and Brazos, 120 cores



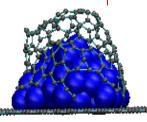
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Nucleation and Growth Mechanisms of Single-Walled Nanotubes on Carbide and Metal Nanoparticles



- Electron density transfer between growing nanotube and catalytic nanoparticle calculated using DFT in VASP[®]
- Simulation consisted of 55 Ni Atoms and 100+ C atoms and was ran in parallel on Ada ~80 CPUs

- Reactive Molecular Dynamics simulation demonstrating the catalytic growth process
- SIMCAT Fortran code
- Code ran in 1 CPU for 600+ hours

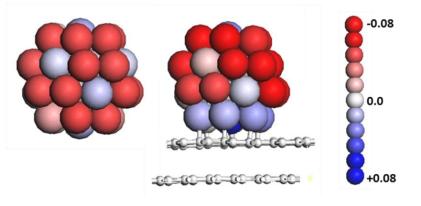




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Catalyst Supported on a Substrate: Charge Transfer

Bader Charges: After adsorption, on the interface the charge of the atoms is positive, on the top is more negative.

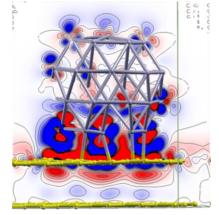


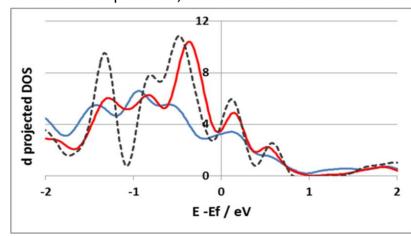
J. Ma, G. Ramos-Sanchez, PB Balbuena, N. Alonso-Vante et al, ACS Catalysis, 3, 1940-1950, (2013)

Details of the electronic transfer during reaction and the effect of the catalyst support (carbon) incorporated to the DFT simulations

Total effect: The high hybridization, modification of the structure and charge transfer lead to more states near the Fermi level of top atoms, the DOS of these atoms are

shifted closer to the Fermi level.





Accomplished	
through simulations	software: VASP;
in	hardware: EOS,
high performance	Stampede
computers!!!	

G. Ramos-Sanchez and P. B. Balbuena, PCCP, 15, 11950-11959, (2013)



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Project: Coke-resistant catalysts for the dry reforming reaction of CH₄

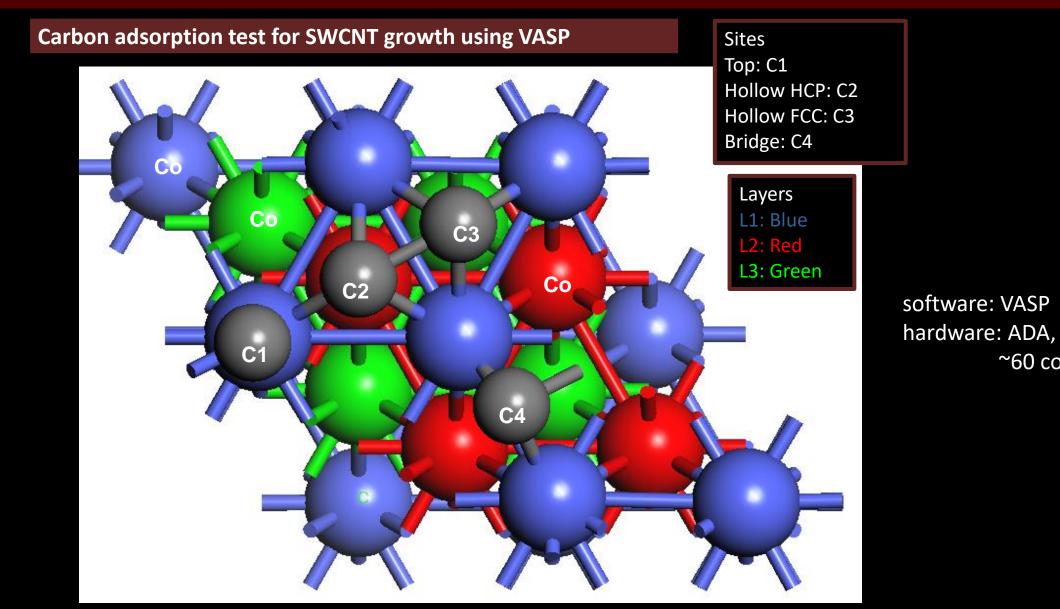
Figure: NiO(111)-*p*(2x2) octopolar surface which exhibits coke resistant property



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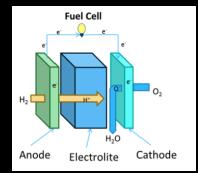
~60 cores

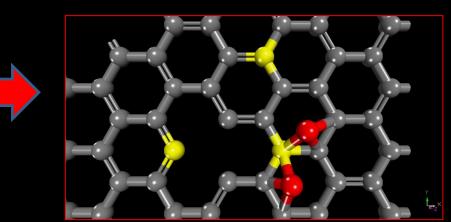




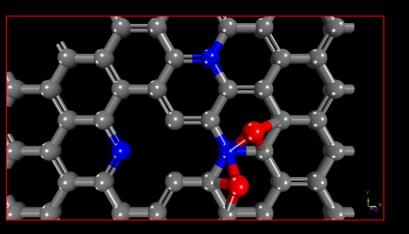
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Modeling Electrochemical Reactions on Nanostructured Carbon Materials





software: VASP hardware: ADA, ~60 cores



NitrogenSulfurOxygen





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HIGH PERFORMANCE RESEARCH COMPUTING TEXAS A&M UNIVERSITY

Brazos HPC Cluster





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